TWO NEW FLAVONOIDS FROM ERYTHRINA ERIOTRIOCHA¹

AUGUSTIN E. NKENGFACK,² DALE R. SANSON, MICHAEL S. TEMPESTA,*

Department of Chemistry, University of Missouri, Columbia, Missouri 65211

and Z. TANEE FOMUM*

Department of Organic Chemistry, University of Yaounde, B.P. 812, Yaounde, Cameroon

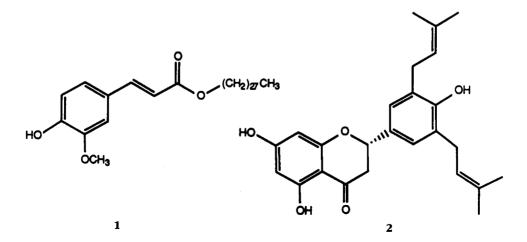
ABSTRACT.—In addition to the known compounds erythinasinate [1] and abyssinone V [2], two new flavonoids were isolated from the CHCl₃ extract of the stem bark of *Erythrina erio-triocha* and characterized by spectroscopic techniques as 3'-prenylnaringenin [3] and 2'-hydroxy-5'-methoxybiochanin A [4].

The genus *Erythrina* (Leguminosae) is well known for its alkaloids whose main physiological property is curare-like action (1). In recent years, however, there has been an increase in research efforts on the neutral biologically active compounds of this genus (2-10). In continuation of our phytochemical studies on the Cameroonian species of the genus (3-9), we have examined the constituents of *Erythrina eriotriocha*, a species which has not been investigated previously. In this paper we report the isolation and structural elucidation of two new compounds: a prenylated flavanone, 3'-prenylnaringenin [3] and a methoxyisoflavone, 2'-hydroxy-5'-methoxybiochanin A [4]. We also report ¹³C-nmr spectral data of the co-occurring erythrinasinate [1].

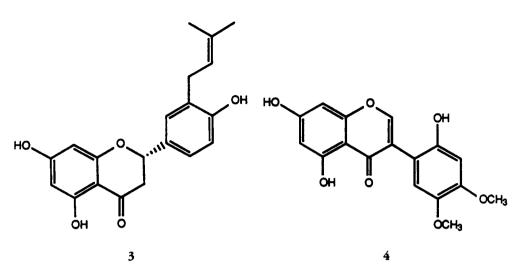
RESULTS AND DISCUSSION

Si gel chromatography of the CHCl₃ extract of *E. eriotriocha* followed by preparative tlc afforded two new flavonoids **3** and **4** along with the previously known erythrinasinate [**1**] (5) and abyssinone V [**2**] (2,9). ¹³C-nmr spectral data [CDCl₃, δ 167.4 (C-3'), 147.9 (C-3), 146.7 (C-4), 144.6 (C-1'), 127.0 (C-1), 123.1 (C-6), 115.7 (C-2), 114.7 (C-5), 109.3 (C-2'), 64.6 (C-1''), 55.9 (OMe), 31.9 (C-26''), 29.7 (C-5'') \rightarrow C-24''), 29.4 (C-4''), 29.3 (C-25''), 28.8 (C-2''), 26.0 (C-3''), 22.7 (C-27''), and 14.1 (C-28'')] of erythrinasinate [**1**] is reported here for the first time.

3'-Prenylnaringenin [3], C₂₀H₂₀O₅ ([M]⁺ 340.1315, calcd 340.1311), [\alpha]²³D



¹Part 13 in the Series "Erythrina Studies." For Part 12 see Phytochemistry, in press. ²On leave from the University of Yaounde.



-1.1 (c = 2.2, MeOH), isolated as an oil, gave positive tests both with FeCl_a (greenish-brown) and Mg-HCl (purple). Its ir spectrum exhibited absorptions at 3410 (free OH), 3240 (chelated OH), and 1640 cm^{-1} (conjugated carbonyl). The uv spectrum in MeOH (227 sh, 290, 312 nm) suggested the presence of a flavanone skeleton. This skeleton was also supported by its ¹³C-nmr spectrum (Table 1) which showed peaks for C-2 and C-3 carbons at δ 79.2 and 42.0, respectively. In the ¹H-nmr spectrum in DMSO- d_6 (Table 1), the downfield signals at δ 12.21, 9.50, and 8.35 confirmed the presence of a chelated hydroxy group at C-5 (δ 12.21) as well as the presence of two free phenolic groups. Two double doublets at δ 2.67 and 3.25 (J = 3.5, 16.4 Hz and 13.8, 16.4 Hz) and a double doublet at δ 5.37 (J = 3.5, 13.8 Hz) were assignable to H_2 -3 and H-2, respectively. The presence of one isoprenyl group was also shown in the ¹H-nmr by two methyl signals (δ 1.69, 1.78), one 2H doublet (δ 3.42), and a 1H triplet (δ 5.25). In addition, a typical ABX system at δ 6.83 (d, J = 8.7 Hz), 7.10 (dd, J = 2.8, 8.7 Hz), and 7.13 (d, J = 2.8 Hz) established the presence of three aromatic protons in ring B with ortho, ortho/meta, and meta coupling, respectively. Also, two 1H doublets with meta coupling (δ 2.1 Hz) at δ 5.84 and 5.86 were consistent with aromatic protons H-8 and H-6 in ring A, respectively. On biogenetic grounds (11), it was assumed that one of the two free phenolic groups is located at C-7 in ring A while the second one is in the B ring at C-4'. The chemical shifts of the B-ring protons were consistent with assignment of the ABX spin systems as H-2', H-5', and H-6', with OH at C-4' and prenyl at C-3'. The isoprenyl group was also shown to be located in the B ring by mass spectral studies. Confirmation of this structure was obtained by the mass spectrum of 3, in which the molecular ion was detected at m/z 340. Fragment ion peaks at m/z 153 and 187 were caused by usual RDA fragmentation followed by hydrogen transfer. The former ion (m/z 153) resulted from the ring-A moiety and showed that this ring possessed two hydroxyls and no isoprenyl. On the other hand, the latter ion $(m/z \ 187)$ arose from the B-ring moiety and showed that the B ring had an isoprenyl as well as an hydroxyl. On the basis of the above spectroscopic studies, structure 3 was assigned to be 5,7,4'-trihydroxy-3'- $(\gamma,\gamma$ -dimethylallyl) flavanone. The absolute stereochemistry at C-2 is assumed to be (S) in accord with known (-)-flavanones (12).

2'-Hydroxy-5'-methoxybiochanin A [4] was obtained as white needles from MeOH/ CH₂Cl₂, mp 250°. Its molecular formula, $C_{17}H_{14}O_7$, was assigned from the hrms, which showed a molecular ion at m/z 330.0736. Its ir spectrum (Nujol) displayed absorptions at 3464 (free hydroxyl), 3246 (chelated hydroxyl), and 1647 cm⁻¹ (conju-

Journal of Natural Products

	Compound					
Atom	3*			4 ^b		
	ιH	J (Hz)	¹³ C	¹ H	J(Hz)	¹³ C
1	_		_		_	
2	5.37,dd	3.5,13.8	79.2	8.02	·	155.5
3	3.25,2.67,dd	3.5, 16.4; 13.8, 16.4	42.0	—	— I	110.4
4		—	196.1	_		181.0
5	_		162.9		_	158.5
6	5.86,d	2.1	95.9	6.43,d	2.7	99.3
7	_	_	163.5		-	164.5
8	5.84,d	2.1	95.1	6.27,d	2.7	94.0
9	_		166.6	—	—	163.3
10		—	101.7	—	-	105.6
1'			128.8	—		121.0
2'	7.13,d	2.8	125.5		—	153
3'		_	127.4	6.64	0	100.8
4'		—	155.3	—	- 1	148.3
5'	6.83,d	8.7	114.6		_	141.5
6'	7.10,dd	2.8,8.7	122.6	6.94	0	116.4
1″	3.42,d	7.1	28.1	—		—
2″	5.25,t	7.1	128.2	—	_	—
3″	_	-	131.5			
4"	1.69,s	_	17.7		—	—
5″	1.78,s	· _	25.6	—		
5-OH	12.21	_		13.02		—
7 -OH	9.50		-	9.98	—	
2'-OH		_		7.95		
4'-OH	8.35		_		—	—
4'-OMe		_		3.80		56.1
5'-OMe	—		_	3.68	-	56.7

 TABLE 1.
 ¹H-nmr Data and ¹³C-nmr Assignments for 3'-Prenylnaringenin [3] and 2'-Hydroxy-5'-Methoxybiochanin A [4].

In DMSO-d₆.

^bIn Me₂CO- d_6 .

gated carbonyl). In the ¹H nmr in Me₂CO- d_6 , a 1H singlet at δ 8.02 was characteristic of an isoflavone and assignable to H-2. This skeleton was also supported by its uv spectrum (295, 266, 259 nm) and color tests (Mg-HCl and FeCl₃). The downfield signals at δ 13.02, 9.98, and 7.95 confirmed the presence of a chelated hydroxyl at C-5 (δ 13.02) as well as the presence of two free phenolics. The presence of two methoxy groups in the ¹H nmr of 4 was also shown by two sharp 3H signals at δ 3.68 and 3.80 ppm. Furthermore, two doublets with meta coupling (J = 2.8 Hz) at δ 6.27 and 6.43 ppm were assigned as H-8 and H-6, respectively. Two sharp 1H singlets at δ 6.64 and 6.94 ppm established the presence of two para protons in the B ring. The mass spectrum of **3** exhibited a molecular ion at m/z 330 and significant fragment peaks resulting from retro-Diels-Alder cleavage of the C ring at m/z 152 and m/z 178, indicating that the two methoxy groups were both located in the B ring. From the above spectroscopic studies and biogenetic considerations (11), three possible structures can be proposed.

To choose the correct structure a 2D nOe experiment (NOESY) was performed (13). We observed an intense nOe between each para proton and one methoxy group (δ 6.64 and 3.80; δ 6.94 and 3.68) as well as an nOe between the two methoxy groups. These results permitted us to assign structure 4 as 2'-hydroxy-5'-methoxybiochanin A.

EXPERIMENTAL

INSTRUMENTAL.—Mass spectra were obtained with a Kratos MS-25 with a DS-55 data system. Ir spectra were run on a Nicolet 20 DBX and uv spectra on a Beckman 25 spectrophotometer. All nmr experiments were performed on a Nicolet NT 300 WB or JEOL-FX 90Q spectrometer equipped with 5-mm ¹H and ¹³C probes operating at 300.06 and 75.45, or 90 and 22.5 MHz, respectively. Samples were run in Me₂CO-d₆, CDCl₃, or DMSO-d₆, and chemical shifts were referenced to internal TMS 0.00 ppm for ¹H nmr and to deuterated solvents for ¹³C-nmr spectra.

PLANT MATERIALS.—*E. eriotricha* stem bark was collected at Meiganga in Adamaoua province of Cameroon in June 1987. Voucher material documenting the collection was identified by the Director of the National Herbarium, Yaounde, Cameroon, and is on deposit there.

EXTRACTION AND ISOLATION.—Dried, ground stem bark (10 kg) was successively extracted in a MeOH Soxhlet extractor with *n*-hexane, CHCl₃, and MeOH. Concentration of extracts under reduced pressure gave, respectively, 60 g (0.6%) of hexane extract and 200 g of CHCl₃ extract. The MeOH extract consisted mainly of tannins. Only the CHCl₃ extract was examined in this investigation. Part of this extract (100 g) was subjected to cc over Si gel (900 g) packed in hexane. Gradient elution was effected with hexane/EtOAc, EtOAc, and MeOH/EtOAc mixtures. A total of 200 fractions (ca. 150 ml per fraction) were collected and mixed on the basis of tlc. The pure compounds were obtained from combined fractions after further purification by cc followed by tlc.

Fractions 5–15, eluted with hexane-EtOAc (10:1) were concentrated to give an oil (100 mg). Preparative tlc on Si gel of this oil with toluene-Me₂CO (10:1) afforded 20 mg of a compound whose mp (76°), ir, uv, and ¹H-nmr spectral data matched with those published (5) for erythrinasinate [1]. Its ¹³C-nmr spectra data is reported for the first time. Fractions 20–30, eluted with hexane-EtOAc (17:3), were concentrated to give a sticky yellow oil (200 mg). Repeated cc of this oil, followed by preparative tlc on Si gel eluted with toluene-Me₂CO (10:2) yielded a compound whose physical and spectroscopic properties were identical with the previously known abyssinone V [2] (2). The combined fractions 15–48, eluted with hexane-EtOAc (10:4), were concentrated to give 100 mg of sticky brown oil. Further cc on Si gel eluted with hexane and increasing concentration of EtOAc yielded 2 fractions, A and B.

Fraction A was rechromatographed by cc on Si gel followed by preparative tlc on Si gel eluted with toluene-Me₂CO (10:3) to yield 7 mg of **3**. Fraction B, eluted with hexane-EtOAc (3:4) was subjected to reversed-phase cc eluted with MeOH-H₂O (10:3), yielding compound **4** (10 mg).

3'-PRENYLNARINGENIN [**3**].—Oil; $[\alpha]^{23}D - 1.1$ (c = 2.2, MeOH); uv λ max (MeOH) (log ϵ) 312 (3.14), 290 (3.24); /AlCl₃ 310 (4.08), 232 (4.46); /NaOMe 323 (4.09), 242 (4.06); ir ν max (CDCl₃) 3410, 3240, 1640, 1215, 1020, 756 cm⁻¹; ¹H nmr (300 MHz, DMSO) see Table 1; ¹³C nmr (75.45, DMSO) see Table 1; eims m/z [M]⁺ 340 (58), 323 (77), 297 (27), 285 (44), 284 (37), 267 (35), 213 (22), 188 (33), 187 (6), 175 (100), 153 (90), 152 (40), 133 (56); hrms calcd for C₂₀H₂₀O₅ m/z [M]⁺ 340.1311, found m/z 340.1315.

2'-HYDROXY-5'-METHOXYBIOCHANIN A [4].—White needles (MeOH/CH₂Cl₂); mp 250°; ir (Nujol) 3464, 3246, 1647, 1627, 1445 cm⁻¹; uv λ max (MeOH) (log ϵ) 295 (3.76), 266 (4.60), 259 (4.75), 253 (4.64); /NaOMe 323 (3.74), 269 (4.72), 260 sh (4.78), 253 (4.64); /AlCl₃ 296 (3.81), 269 (4.69), 263 (4.77), 253 (4.62); /NaOAc 296 (3.88), 268 (4.60), 260 (4.72), 252 (4.60); ¹H nmr and ¹³C nmr see Table 1; eims *m*/z [M]⁺ 330 (100), 315 (36), 300 (48), 178 (8), 177 (12), 168 (8), 153 (44), 152 (14), 135 (22); hrms calcd for C₁₇H₁₄O₇ *m*/z [M]⁺ 330.0734, found *m*/z 330.0736.

ACKNOWLEDGMENTS

We are grateful for partial financial support to the International Foundation for Science (IFS) Research Grant (F/1392-1) and the University of Yaounde, Cameroon Research Grants Committee, to the National Science Foundation for the 300 MHz NMR Spectrometer (PCM-8115599) and the Kratos MS-25 (PCM-8117116) Instrument, and to the University of Missouri Institutional Biomedical Research Support Grant (RR07053) from the National Institutes of Health.

LITERATURE CITED

- S.F. Dyke and D. Quessy, in: "The Alkaloids, Chemistry and Physiology." Ed. by R.G.A. Rodrigo, Academic Press, New York, 1981, Vol. 18, p. 1.
- 2. S.V. Kamat, Y.F. Chuo, I. Kubo, and K. Nakanishi, Heterocycles, 15, 1163 (1981).
- 3. Z.T. Fomum, J.F. Ayafor, and J.T. Mbafor, Tetrabedron Lett., 24, 4127 (1983).
- R.B. Taylor, M.S. Tempesta, Z.T. Fomum, J.F. Ayafor, and J. Wandji, J. Nat. Prod., 49, 670 (1986).

- 5. Z.T. Fomum, J.F. Ayafor, A.E. Nkengfack, J. Wandji, and W.G. Fomban, Phytochemistry, 25, 757 (1986).
- Z.T. Fomum, J.F. Ayafor, P.N. Ifeadike, A.E. Nkengfack, and J. Wandji, Planta Med., 4, 341 (1986).
- 7. R. Promsattha, M.S. Tempesta, Z.T. Fomum, J.F. Ayafor, and J.T. Mbafor, J. Nat. Prod., 49, 932 (1988).
- 8. Z.T. Fomum, J.F. Ayafor, J.T. Mbafor, and C.N. Mbi, J. Chem. Soc., Perkin Trans. 1, 33 (1986).
- 9. R. Promsattha, M.S. Tempesta, Z.T. Fomum, and J.T. Mbafor, J. Nat. Prod., 51, 611 (1988).
- L.A. Mitscher, S.R. Gollapudi, D.C. Gerlach, S.D. Drake, E.A. Veliz, and A.J. Ward, Phytochemistry, 27, 381 (1987).
- 11. E. Ebel and K. Hahlbrock, in: "The Flavonoids—Advances in Research." Ed. by J.B. Harborne and T.J. Mabry, Chapman and Hall, New York, 1982, p. 649.
- 12. W.B. Whalley, in: "The Chemistry of Flavonoid Compounds." Ed. by T.A. Geissman, Pergamon, London, 1962, p. 441.
- 13. J. Jeener, B.H. Meier, P. Bachman, and R.R. Ernst, J. Chem. Phys., 71, 4546 (1979).

Received 12 October 1988